Water Rennovation in Ukraine

Project no. 22320101





Water Rennovation in Ukraine

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The project is co-financed by the Governments of the Czechia, Hungary, Poland and Slovakia through Visegrad Grants from International Visegrad Fund. The mission of the fund is to advance ideas for sustainable regional cooperation in Central Europe.

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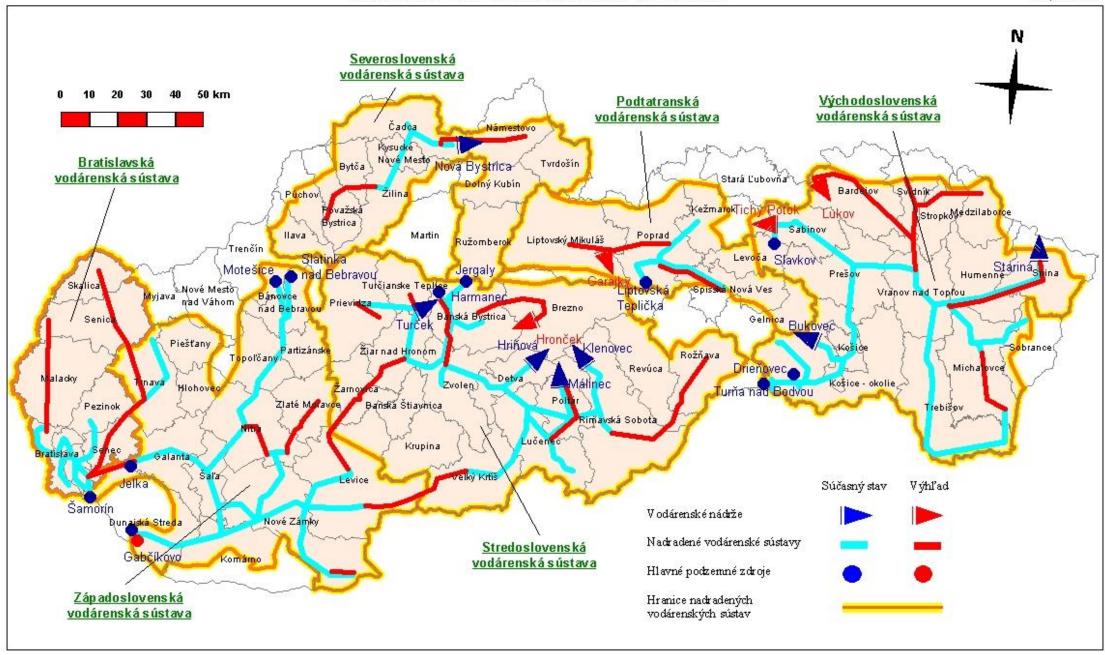




WWTP for the urban areas

Lubos Jurik

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Composition of Wastewater

Visegrad Fund

- Inorganics
 - Ammonia
 - Nitrate
 - Phosphate
 - Carbonate
 - Minerals
 - Calcium
 - Magnesium
 - Iron
 - Etc.

Organics

- Biodegradable (BOD)
 - Carbohydrates
 - Proteins (TKN)
 - FOG
- Non-Biodegradable (COD-BOD)
 - Large particles
 - Complex polymers (plastics, lignin)
 - Surfactants (some)
 - Pesticides (some)
 - Pharmaceuticals (some)

Bacteria

- $C_5H_7O_2N$
- Growth rates double every 10°C until optimum temp reached
- pH optimum 6.5-7.5
- Nutrients may be limited in industrial wastewaters

Constituent or element	Percent of dry weight	•
Major cellular material		1.5
Protein	55.0	segrad Fund
Polysaccharide	5.0	
Lipid	9.1	
DNA	3.1	
RNA	20.5	
Other (sugars, amino acids)	6.3	
Inorganic ions	1.0	
As cell elements		
Carbon	50.0	
Oxygen	22.0	
Nitrogen	12.0	
Hydrogen	9.0	
Phosphorus	2.0	
Sulfur	1.0	
Potassium	1.0	
Sodium	1.0	
Calcium	0.5	
Magnesium	0.5	
Chlorine	0.5	
Iron	0.2	5
Other trace elements	0.3	5

Theoretical Yield [(g cell / g glucose) / (g COD O2 / g glucose)]



Solve for the theoretical yield

$$3C_6H_{12}O_6 + 8O_2 + 2NH_3 \rightarrow 2C_5H_7NO_2 + 8CO_2 + 14H_2O_3$$

3(180) 8(32) 2(17) 2(113)

COD =
$$\frac{\Delta(O_2)}{\Delta(C_6H_{12}O_6)} = \frac{6(32 \text{ g/mole})}{(180 \text{ g/mole})} = 1.07 \text{ g O}_2/\text{g glucose}$$

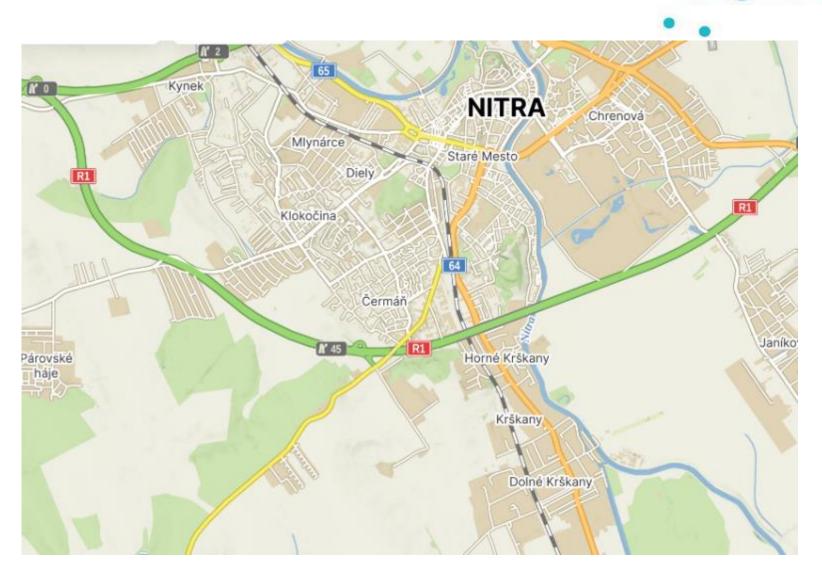
$$Y = \frac{\Delta (C_5 H_7 NO_2)}{\Delta (C_6 H_{12} O_6 \text{ as COD})} = \frac{2(113 \text{ g/mole})}{3(180 \text{ g/mole})(1.07 \text{ g COD/g glucose})}$$

= 0.39 g cells/g COD used

WWTP Nitra



WWTP Nitra location



WWTP Nitra

Visegrad Fund

Project ISPA: 2000/SK/16/P/PE/002

Financing:

ISPA/kohézny fond/ 50% 5 177 100 EUR

Slovakia 25% 2 588 550 EUR

Nitra Town 25% 2 588 550 EUR

Total: 10 354 200 EUR

Investment: Mesto Nitra

Technical advisor: Carl Bro Group, Denmark

Costruction: Conzortium VÁHOSTAV-KPRIA- INGSTAV

Buiding parts: Váhostav – SK, a.s. Žilina

Technologie: Královopolská a.s, Ingstav a.s.

User: Town Nitra and Západoslov. vodárenská spoločnosť, a.s. Nitra

WWTP Nitra location

- Visegrad Fund
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WWTP Nitra

Visegrad Fund

Finished: 18.10.2007

WWTP Nitra Krškany is placed at left bank of Nitry river.

Capacity of WWTP is 212 000 El. Yearly outflow is 13 249 500 m³

WWTP Nitra



The Wastewater Treatment Process

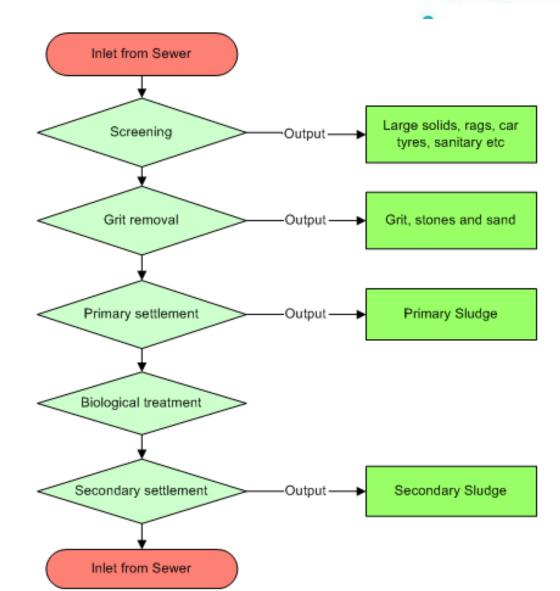
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Preliminary Treatment

Primary Treatment

Secondary Treatment

Tertialy treatnent



Inflow part



Inflow from the municipality



Flow measurement at the inflow



Gravel trap at the inflow before pumpingsegrad Fund



Coarse screens



Captured grabs at screens







Mechanical pretreatment

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Pumping station Archchimedes screew



Pumping station Archchimedes screew



Outflow from fine screens



Sand trap and fats and oils captures





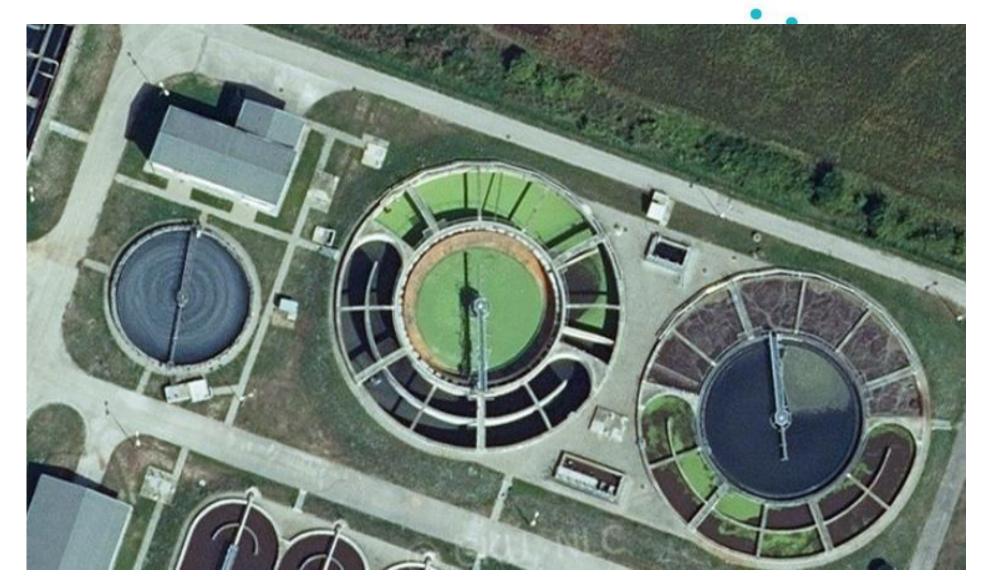


Sand trap





Primary settlement and Sludge regeneration and Fund



Primary setlement





Primary setlement





Primary setlement





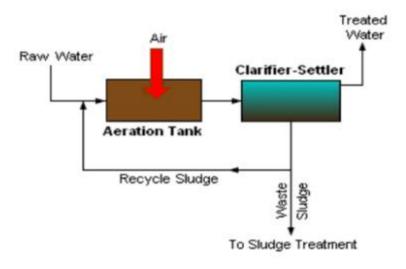


Outflow from primary settlement to biological treatment



Activated Sludge (ASP)

Activated sludge consists of a mass of micro-organisms which feed on pollutants in the sewage. The bacteria is suspended in liquid and is called "mixed liquor", it is mixed with sewage and aerated in aeration basins before passing to final settlement tanks where it is settled and the sludge returned. The effluent produced is of a high quality.





Biological treatment



Biological Treatment





Activated Sludge technology - construction rad Fund



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ad Fund

Air blowers BOD5





Nitrification

- $NH_4^+ \rightarrow Nitrosomonas \rightarrow NO_2^-$
- $NO_2^- \rightarrow Nitrobacter \rightarrow NO_3^-$
- Notes:
 - Aerobic process
 - Control by SRT (4 + days)
 - Uses oxygen \rightarrow 1 mg of NH₄⁺ uses 4.6 mg O₂
 - Depletes alkalinity → 1 mg NH₄⁺ consumes 7.14 mg alkalinity
 - Low oxygen and temperature = difficult to operate

Denitrification

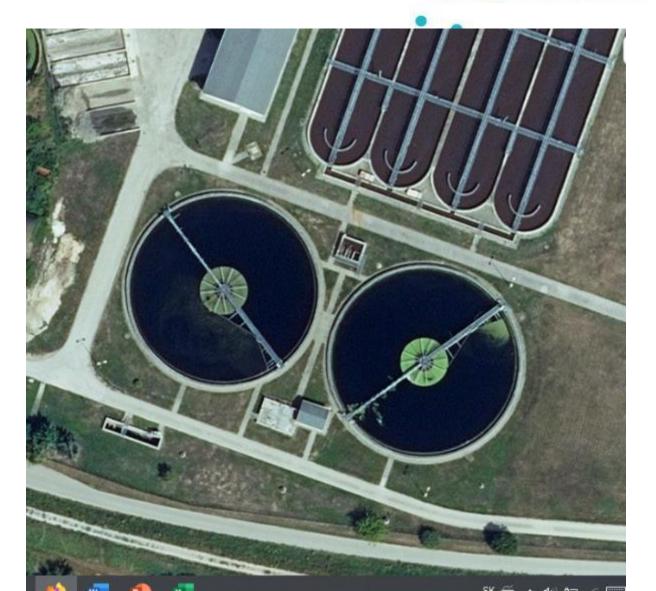


Using methanol as carbon source:

$$6 \text{ NO}_3^- + 5 \text{ CH}_3 \text{OH} \qquad -\text{N}_2 + 5 \text{ CO}_2 + 7 \text{ H}_2 \text{O} + 6 \text{ OH}^-$$

Using an endogenous carbon source:

Finaly settlement



Finaly setlement



WWTP finaly setlement



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Sludge regeneration



Sludge treatment



Sludge treatment



WHAT IS SLUDGE?

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Bacteria and other micro-organisms

Mainly water (up to 99%)
Dissolved solids
Settled and suspended solids

Nutrients (N, P, K)

Faecal matter

Metals

Energy



Sewage sludge

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WWTP Nitra Sluge stabilisation and Biogas grad Fund







Sludge stabilisation



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Biogas storage space



Incineration of surplus gas



Outflow to the river

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